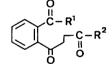
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SYNTHESIS AND REVISED STRUCTURE OF THE O-SUCCINYLBENZOIC ACID COENZYME A ESTER, AN INTERMEDIATE IN MENAQUINONE BIOSYNTHESIS

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Summary: Synthesis of the two isomers 2, 3 of the mono coenzyme A ester of o-succinylbenzoic acid (1, OSB, i.e. 4-(2-carboxyphenyl)-4-oxobutanoic acid) and enzymic conversion of 3 to 1,4-dihydroxy-2-naphthoic acid Z shows that as opposed to previous assumptions the "aliphatic" rather than the "aromatic" carboxyl group in o-succinylbenzoic acid 1 is activated during vitamin K_2 biosynthesis in Escherichia coli and Mycobacterium phlei.

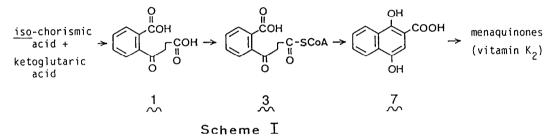
The enzymic conversion of o-succinylbenzoic acid 1 to 1,4-dihydroxy-2-naphthoic acid (DHNA) 7 is an ATP, CoASH and Mg^{2+} dependent reaction¹. The product DHNA 7 is a precursor of vitamin K₂ (menaquinone). The activated intermediate in the conversion of OSB 1 to DHNA 7 has been isolated, characterized and shown to be a mono coenzyme A ester of OSB^{1b}.



 $\begin{array}{cccc} 1 & R^{1}, R^{2} = OH & & & & & & \\ 2 & R^{1} = CoAS, R^{2} = OH & & & & & & \\ 3 & R^{1} = OH, R^{2} = CoAS & & & & & & \\ 8 & R^{1} = OCH_{3}, R^{2} = OH \end{array}$

Since OSB <u>1</u> has two carboxyl groups the question arose which of the two carboxyl groups in OSB <u>1</u> is activated. This question was answered by synthesis of both isomers <u>2</u>, <u>3</u> and enzymic conversion^{1b} of <u>3</u> but not <u>2</u> to DHNA <u>7</u>: OSB <u>1</u> was treated with 1,1'-carbonyldiimidazole in a molar ratio of 2,2 to 1. The resulting mixture of OSB monoimidazolides <u>4</u>, <u>5</u> was converted to a mixture of coenzyme A esters² <u>2</u>, <u>3</u>. The coenzyme A esters <u>2</u>, <u>3</u> were separated by HPLC. On hydrolysis both esters gave OSB <u>1</u> and coenzyme A in a molar 1 to 1 ratio. The "aromatic" ester <u>2</u> was distinguished from the "aliphatic" ester 3 by a synthesis which gave the "aromatic" ester 2 alone: Treatment of OSB 1 with an excess of 1,1'-carbonyldiimidazole gave OSB diimidazolide 6. Mild acid hydrolysis (THF/acetic acid 1:1, pH 3,2, 30°C, 30 min) gave 4 which was characterized (elementary analysis, UV, IR, MS, ¹H NMR, derivatisation to the corresponding methylester <u>8</u>). The imidazolide 4 was also converted 2 to the corresponding coenzyme A ester 2.

When the "aromatic" 2 and the "aliphatic" 3 coenzyme A ester were incubated with the previously used naphthoate synthase fraction^{1b}, the "aliphatic" coenzyme A ester 3 was converted to DHNA 7 in a 50 - 60% yield without cofactor requirement. The "aromatic" coenzyme A ester 2, however, was not converted. Therefore menaquinone biosynthesis takes place as shown in Scheme I:



This result invalidates previous assumptions^{1a,b} and the interpretations of experiments 4,5 which led to the conclusion that 2 rather than 3 is the intermediate in vitamin K^4 and anthraquinone⁵ biosynthesis. As will be shown elsewhere⁶ this error was eventually traced back to the fact that the enzymically formed^{1b} OSB mono coenzyme A ester consisted of both the "aliphatic" 3 (85 %) as well as the "aromatic" 2 (15 %) coenzyme A ester and the apparent instability of 3 as opposed to 2.

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